Nanostructure rod-like TiO₂-reduced graphene oxide composite aerogels for highly-efficient visible-light photocatalytic CO₂ reduction

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Abstract

In response to the worldwide over-standard carbon dioxide emission problem ^[1, 2], this work synthesized a series of titanium dioxide/reduced graphene oxide composite aerogels (rGO-Ti) for photoconversion of CO₂ by a one-step hydrothermal and freezedrying method. The microstructure, chemical composition, optical characteristics and photocatalytic reduction activity of samples have been tested and analyzed. The conclusions were summarized as follows: (1) The 3-D network structure rGO aerogel are introduced to rod-like TiO₂ with a diameter of 10-20 nm and length of 100-150 nm. The hydrothermal reaction between Ti⁴⁺ and graphene formed strong O=C-O-Ti group which is related to specific chemical bonding sites. (2) The surface area of rGO-25Ti aerogel is as high as 287.3 m²/g and it also has high pore volume of 0.72 cm³/g, which is beneficial for more absorption of reactants and rapid transfer of products. (3) Under visible light irradiation, the rGO reinforced samples, especially in-situ rGO-Ti aerogel, exhibit much higher light absorption efficiency than pure P25, in which the graphene sheets play an important role in accelerating electron transfer. (4) According the Kubelka-Munk function, the calculated band gap value of rGO-25Ti (TiO2-rGO with 25mmol Ti⁴⁺, 2.9eV) is lower than that of commercial P25 (3.2eV), resulting in red shift in the absorption edge and contributing higher photocatalytic activity. (5) When the addition amount of TiCl₄ to graphene solution for hydrothermal was 25 mmol, the highest carbon conversion of TiO2-rGO aerogel was 21.38 µmol/g, as much as 15.7 times of the conversion rate of pure P25. To sum up, the rGO aerogel-based photocatalyst composite synthesized with TiCl₄ as precursor was applied to photocatalytic CO₂ reduction for the first time. This work indicates that this sort of material could be used as a promising photocatalyst for the photoreduction of carbon dioxide ^[3, 4].

Keywords

Rod-like TiO₂; reduced graphene oxide; aerogel; nano composite; visible light photocatalysis; CO₂ conversion

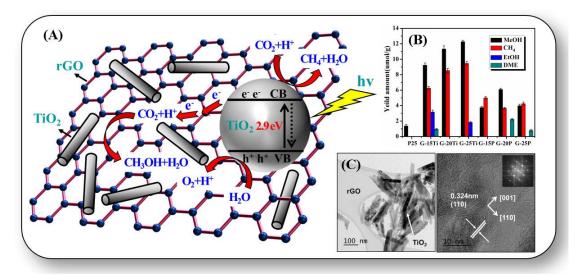


Fig.1 (A) Schematic illustration of the band structure and charge separation in TiO2rGO, (B) MeOH, CH₄, EtOH and DME yields amount of photocatalytic conversion of CO₂ for different catalysts and (C) TEM micrographs of G-25Ti composite.

Reference

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